A time dependent crack growth law for high temperature conditions

- C. Timbrell¹, R. Chandwani¹, D.W. MacLachlan², S.J. Williams²
- 1. Zentech International Limited, 590b Finchley Road, London, NW11 7RX, UK
 - 2. Rolls-Royce plc, Elton Road, Derby, PO Box 31, DE24 8BJ, UK

1 Introduction

Alloys, especially nickel based ones used in the aerospace industry, are continuously being improved to provide greater strength against component failure and also to increase resistance against crack propagation. This involves altering their composition and, under controlled conditions, modification of precipitate and grain sizes. At high temperatures under both sustained and cyclic loading conditions, these microstructural changes interact synergistically with time dependent mechanisms such as creep, oxidation and corrosion and affect the crack growth rate (CGR). The individual effects of environmental conditions such as oxidation and corrosion and microstructural evolution of grain size at high temperatures, are generally difficult to evaluate. In addition, thermo-mechanical testing of large numbers of specimens under a variety of conditions can be prohibitively costly. Attempts have been made over the last few decades by a number of investigators to conduct standardised tests under controlled environmental conditions and compare them with the results obtained in neutral environments such as vacuum or inert gas [1-4]. It has been found that these environmental effects interact and their combined effect is generally greater than if they were considered separately. In this paper a time dependent crack growth law, COMET (Creep Oxidation Microstructure Environment Temperature), is described which considers the effect of these combined processes using a temperature dependent parameter based on an Arrhenius equation. Using this time dependent law in conjunction with a fatigue crack growth law, a finite element based implementation has been developed to carry out detailed 3D crack propagation analysis and simulation of a cracked component under the effect of thermo-mechanical loading at high temperatures.

2 COMET crack growth law

The COMET equation for time dependent crack growth data defines the instantaneous time dependent crack growth rate as:

$$\frac{da}{dt} = D(K)^n$$
 Equation 1

where:

a = crack size D, n = material parameters (see below)

t = time K = stress intensity factor

3 Definition of the D parameter

The time dependent mechanisms that are active during the time the component remains at high temperature (also referred to as "hold" or "dwell" time), may include creep, oxidation, corrosion and microstructural evolution of grain size and structures. These time dependent effects interact synergistically and their combined effect is generally greater than if they were considered separately. Also it must be noted that it is not easy or cost effective to determine these effects individually. For example, during the dwell period it is not clear whether the increase in crack growth rate is purely due to creep or due to the environmental interaction involving oxidation and corrosion. The time dependent growth is assumed only to occur when the applied stresses are tensile. Therefore the term D in Equation 1 is evaluated only when K is greater than zero. This is because the crack will be closed when subject to compressive loads and therefore the effects of creep and oxidation will be significantly reduced.

The parameter, D, used to describe time dependent growth rates is temperature dependent. Rate based temperature dependent quantities are frequently defined to vary with temperature based on an Arrhenius equation:

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$$D = A e^{\left(\frac{-Q}{RT}\right)}$$
 which can be written as

 $D = A \ e^{\left(\frac{-B}{T}\right)}$

Equation 2

where :

A, B are material constants R is Universal Gas Constant

Q is Activation Energy T is temperature in degrees Kelvin

Q is based on the activation barrier for the synergistic mechanism affecting crack growth. A and B are temperature independent material constants that can be determined experimentally by carrying out isothermal specimen testing at a range of temperatures of interest and preferably with different dwell periods.

4 Calculation of material constants A and B

One methodology for the evaluation of the values of A and B is *summarised* below. This procedure first requires determination of values of the Arrhenius parameter, D, for a range of temperatures.

- At several temperatures, T, over a range of temperatures of interest, crack growth data is
 obtained experimentally using a trapezoidal waveform loading cycle consisting of ramp up,
 hold at max, ramp down and hold at min. We refer to the duration of each stage of this cycle
 using the notation 1.1.1.1, for instance where each stage lasts 1 second. Varying dwells at
 max load of between 1 second to 3600 seconds or even longer may be used ie 1.3600.1.1.
 The 1.1.1.1 cycle is referred to as a baseline cycle, due to the absence of a significant dwell
 this cycle is used to define the time independent response and is assumed not to contain any
 time dependent crack growth.
- For each temperature (T) plot the curve (log-log) of da/dN vs ∆K which has the contribution from both time independent and time dependent CGRs.
- To obtain only time dependent crack growth data at each temperature, subtract the baseline fatigue crack growth rate from the overall crack growth rate for the tests with a max load dwell time greater than one second. Fit this new data to Equation 1 using a log-log scale to get a graph of log (da/dt) vs log (K) for each temperature.
- From each isothermal curve, extrapolate to K=1.0MPam^{1/2}, obtain the value of da/dt in ms⁻¹ which is equal to D for that temperature. (Note that different specimen tests under the same nominal test conditions will produce scatter and therefore a range of D values can be derived at each temperature. In the following, a single value of D is used for each temperature).
- Thus for each value of temperature (T), we have a value of D from these fitted curves.
- Now to obtain the values of material constants A and B, consider the Arrhenius relationship in Equation 2:

$$D = A e^{\left(\frac{-B}{T}\right)}$$
 which gives: $\ln(D) = \ln(A) - \frac{B}{T}$ Equation 3

• On a In-linear scale plot D vs 1/T where T is in degrees Kelvin to fit Equation 3. The intercept of this curve gives the value of A and the slope of this curve is –B.

5 Threshold temperature and blend function

Equation 1 is valid at high temperature i.e. above a threshold temperature, T_{thr} . Below the threshold temperature the value of D (and hence the time dependent crack growth rate) is defined as zero. To provide a smooth transition across the threshold temperature a value T_{blend} is defined such that in the range T_{thr} to ($T_{thr}+T_{blend}$), Equation 3 is replaced by an alternative definition. The full definition of D as a function of temperature is then:

$$D = 0 for T \le T_{thr} Equation 4$$

$$D = E \left(T - T_{thr}\right)^F for T_{thr} < T < T_{thr} + T_{blend} Equation 5$$

$$D = A e^{\left(\frac{D}{T}\right)}$$
 for $T_{thr} + T_{blend} \le T$ Equation 6

where E and F are determined by the requirement to match the values of D and dD/dT from Equation 3 at a temperature T_{blend} above the threshold.

6 Acknowledgements

The authors gratefully acknowledge the support of the UK Technology Strategy Board for this research project DISPLACE – Developing Improved Service Propagation Lives in Arduous Cyclic Environment (TP/8/MAT/6/I/Q1525K).

7 References

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